

Héctor Novoa de Armas,<sup>a\*</sup>  
Oswald M. Peeters,<sup>a</sup> Norbert M.  
Blaton,<sup>a</sup> Camiel J. De Ranter,<sup>a</sup>  
Margarita Suárez Navarro,<sup>b</sup>  
Esperanza Salfrán Solano,<sup>b</sup>  
Yamila Verdecia Reyes<sup>b</sup> and  
Estael Ochoa Rodríguez<sup>b</sup>

<sup>a</sup>Laboratorium voor Analytische Chemie en Medicinale Fysicochemie, Faculteit Farmaceutische Wetenschappen, Katholieke Universiteit Leuven, Van Evenstraat 4, B-3000 Leuven, Belgium, and <sup>b</sup>Laboratorio de Síntesis Orgánica, Facultad de Química, Universidad de La Habana, Apartado 10400 La Habana, Cuba

Correspondence e-mail:  
hector.novoa@pharm.kuleuven.ac.be

#### Key indicators

Single-crystal X-ray study  
T = 293 K  
Mean  $\sigma(C-C)$  = 0.006 Å  
R factor = 0.055  
wR factor = 0.167  
Data-to-parameter ratio = 12.5

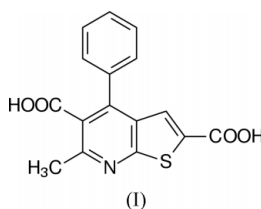
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## 6-Methyl-4-phenylthieno[2,3-*b*]pyridine-2,5-dicarboxylic acid

The crystal structure of the title compound, C<sub>16</sub>H<sub>11</sub>NO<sub>4</sub>S, is stabilized by intermolecular hydrogen bonds of the types O—H···O, O—H···N and C—H···O. The thieno[2,3-*b*]pyridine moiety is planar.

#### Comment

Thieno[2,3-*b*]pyridine derivatives form a class of fused heterocyclic compounds with interesting bio- and pharmacological properties. The activity of some dihydrothieno[2,3-*b*]pyridine derivatives has been thoroughly investigated as calcium antagonists in the treatment of cardiovascular diseases (Dessy *et al.*, 1993). Also, the enantioselective synthesis of a thieno[2,3-*b*]pyridine as a 5-lipoxygenase (5-LO) inhibitor has been reported (Rohloff *et al.*, 1994). In previous work, we described the crystal structures of two different dihydrothieno[2,3-*b*]pyridines (Duque *et al.*, 1998, 2000) and recently we reported the crystal structure of 2,5-diethoxycarbonyl-6-methyl-4-phenylthieno[2,3-*b*]pyridine (Novoa de Armas *et al.*, 2003a) and its substituted 4-(4-bromophenyl) analog (Novoa de Armas *et al.*, 2003b). Saponification of the two ester groups yields the title compound, (I), as a crystalline solid. In (I), the thieno[2,3-*b*]pyridine moiety is planar and the dihedral angle between the least-squares plane of the thieno[2,3-*b*]pyridine moiety and the 4-phenyl ring is 46.3 (2)°. The carboxyl group C21/O22/O23 is nearly coplanar with the thieno[2,3-*b*]pyridine moiety [C3—C2—C21—O22 = 3.2 (7)°], while the other carboxyl group, C51/O52/O53, is not [C4—C5—C51—O52 = 71.2 (6)°]. The mean  $C_{sp^2}-C_{sp^2}$  bond length within the 4-phenyl ring is 1.383 (4) Å. The crystal structure is stabilized by hydrogen bonds of the types O—H···O, O—H···N and C—H···O (Table 2). The difference between the ester 2,5-diethoxycarbonyl-6-methyl-4-phenylthieno[2,3-*b*]pyridine and (I) is the presence of the intermolecular O—H···O hydrogen bonds in the latter.



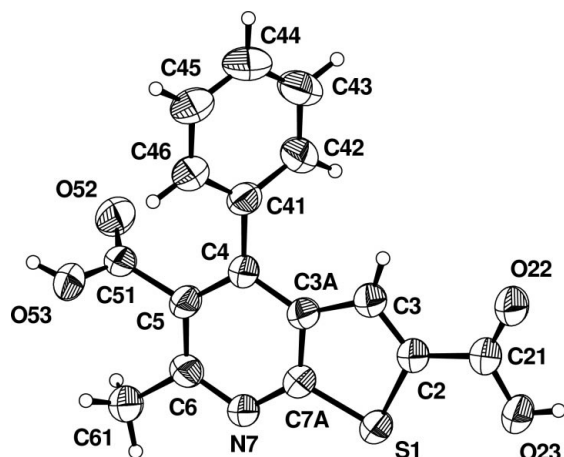
#### Experimental

Diethyl 4-phenylthieno[2,3-*b*]pyridine-2,5-dicarboxylate (2 mmol) was mixed with 20 ml of 10% aqueous sodium hydroxide and the mixture was refluxed vigorously until all the solid ester had dissolved

Received 3 March 2003

Accepted 7 March 2003

Online 14 March 2003



**Figure 1**  
Plot showing the atomic numbering scheme for the title compound. Displacement ellipsoids are drawn at the 50% probability level for non-H atoms.

(about 1 h). The saponification mixture was then cooled and the aqueous solution was separated from any oil present. Afterwards, the solution was acidified to Congo red paper with 10% sulfuric acid. Compound (I) was collected by filtration and washed thoroughly with water, then dried (yield 86%; m.p. 638 K). IR (KBr,  $\text{cm}^{-1}$ ): 3384 (OH), 1724 (C=O), 1685 (C=O), 1600 (C=C);  $^1\text{H}$  NMR (DMSO- $d_6$ , p.p.m.): 7.58–7.51 (*m*, 5H, Ph), 7.49 (*s*, 1H, =CH), 3.44 (*br s*, 2H, OH), 2.66 (*s*, 3H,  $\text{CH}_3$ );  $^{13}\text{C}$  NMR (DMSO- $d_6$ , p.p.m.): 168.6 (CO), 162.2 (CO), 161.1 (C6), 154.2 (C4), 143.0 (C1'), 134.9 (C2), 133.8 (C7a), 128.9 (C4'), 128.6 (C3a), 128.4 (C3' and C5'), 128.3 (C2' and C6'), 127.1 (C3), 126.4 (C5), 22.7 ( $\text{CH}_3$ ); MS, *m/z* (intensity %): 313 ( $M^+$ , 100), 296 (30), 295 (50), 222 (12), 78 (19), 63 (38). Crystals suitable for X-ray analysis were obtained by slow evaporation from methanol.

#### Crystal data

$\text{C}_{16}\text{H}_{11}\text{NO}_4\text{S}$   
 $M_r = 313.32$   
 Monoclinic,  $P2_1/c$   
 $a = 13.3208$  (5) Å  
 $b = 12.5015$  (4) Å  
 $c = 8.9448$  (4) Å  
 $\beta = 102.859$  (3)°  
 $V = 1452.22$  (8) Å $^3$   
 $Z = 4$   
 $D_x = 1.433$  Mg  $\text{m}^{-3}$   
 Cu  $K\alpha$  radiation  
 Cell parameters from 39 reflections  
 $\theta = 6.8$ – $27.9^\circ$   
 $\mu = 2.15$   $\text{mm}^{-1}$   
 $T = 293$  K  
 Plate, colourless  
 $0.40 \times 0.20 \times 0.02$  mm

#### Data collection

Siemens P4 four-circle diffractometer  
 $\omega/2\theta$  scans  
 Absorption correction:  $\psi$  scan (North *et al.*, 1968)  
 $T_{\min} = 0.627$ ,  $T_{\max} = 0.958$   
 3514 measured reflections  
 2535 independent reflections  
 1565 reflections with  $F^2 > 2\sigma(F^2)$   
 $R_{\text{int}} = 0.042$   
 $\theta_{\text{max}} = 69.3^\circ$   
 $h = -16 \rightarrow 16$   
 $k = -15 \rightarrow 1$   
 $l = -1 \rightarrow 10$   
 3 standard reflections every 100 reflections  
 intensity decay: 2%

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.055$   
 $wR(F^2) = 0.167$   
 $S = 1.11$   
 2535 reflections  
 203 parameters  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0433P)^2 + 2.354P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 0.33$  e Å $^{-3}$   
 $\Delta\rho_{\text{min}} = -0.27$  e Å $^{-3}$   
 Extinction correction: SHELXL97  
 Extinction coefficient: 0.0050 (5)

**Table 1**

Selected geometric parameters (Å, °).

S1—C2	1.732 (4)	O52—C51	1.203 (5)
S1—C7A	1.730 (4)	O53—C51	1.322 (5)
O22—C21	1.231 (5)	N7—C6	1.343 (5)
O23—C21	1.309 (5)	N7—C7A	1.341 (5)
C2—S1—C7A	89.7 (2)	N7—C7A—C3A	125.4 (4)
C6—N7—C7A	117.1 (3)	O23—C21—C2	113.8 (4)
S1—C2—C21	121.2 (3)	O22—C21—C2	121.6 (3)
S1—C2—C3	114.2 (3)	O22—C21—O23	124.6 (4)
N7—C6—C61	115.6 (3)	O53—C51—C5	112.5 (3)
N7—C6—C5	121.8 (4)	O52—C51—O53	124.8 (4)
S1—C7A—C3A	113.0 (3)	O52—C51—C5	122.7 (4)
S1—C7A—N7	121.6 (3)		

**Table 2**

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O23—H23...O22 <sup>i</sup>	0.82	1.81	2.633 (5)	176
O53—H53...N7 <sup>ii</sup>	0.82	1.88	2.688 (4)	170
C45—H45...O52 <sup>iii</sup>	0.93	2.48	3.362 (6)	158

Symmetry codes: (i)  $1 - x, 1 - y, 2 - z$ ; (ii)  $2 - x, y - \frac{1}{2}, \frac{3}{2} - z$ ; (iii)  $x, -\frac{1}{2} - y, \frac{1}{2} + z$ .

H atoms were positioned geometrically and included in the refinement, but were constrained to ride on their parent atoms, with  $U_{\text{iso}}(\text{H})$  values fixed at  $1.3U_{\text{eq}}$  of their parent atoms.

Data collection: XSCANS (Siemens, 1996); cell refinement: XSCANS; data reduction: XSCANS; program(s) used to solve structure: SIR92 (Altomare *et al.*, 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: DIAMOND (Bergerhoff, 1996); software used to prepare material for publication: PLATON (Spek, 2003), PARST (Nardelli, 1995) and PARSTCIF (Nardelli, 1991).

HNdA thanks the KU Leuven (Belgium) for support through IRO Scholarships. The staff of the Laboratorio de Síntesis Orgánica, University of Havana (Cuba), is grateful to the Alma Mater projects of that university.

#### References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.  
 Bergerhoff, G. (1996). *DIAMOND*. Gerhard-Domagk-Str. 1, 53121 Bonn, Germany.  
 Dessy, Ch., Salomone, S., Morel, N. & Godfraind, T. (1993). *Eur. J. Pharmacol.* **231**, 435–439.  
 Duque, J., Pomés, R., Díaz, G., Roque, E., Suárez M., Verdecia Y., Ochoa E., Pita, B., Espinosa, R. & Alba, L. (1998). *Acta Cryst.* **C54**, IUC9800002.  
 Duque, J., Pomés, R., Suárez, M., Ochoa E., Verdecia Y., Punte G. & Echevarria, G. (2000). *Z. Kristallogr. New Cryst. Struct.* **215**, 361–362.  
 Nardelli, M. (1991). *PARSTCIF*. University of Parma, Italy.  
 Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.  
 North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.  
 Novoa de Armas, H., Peeters, O., Blaton, N., De Ranter, C., Salfrán Solano, E., Suárez Navarro, M., Ochoa Rodríguez, E. & Verdecia Reyes, Y. (2003a). *Acta Cryst.* **E59**, o321–o323.  
 Novoa de Armas, H., Peeters, O., Blaton, N., De Ranter, C., Salfrán Solano, E., Suárez Navarro, M., Ochoa Rodríguez, E. & Verdecia Reyes, Y. (2003b). *Acta Cryst.* **E59**, o384–o386.

Rohloff, J. C., Alfredson, T. V. & Schwartz, M. A. (1994). *Tetrahedron Lett.* **35**, 1011–1014.  
Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.

Siemens (1996). *XSCANS*. Version 2.2. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.